

PACS numbers: 72.80.Tm, 77.22.Ch, 77.22.Gm, 78.20.Ci, 81.07.Pr, 81.40.Tv, 82.35.Np

Fabrication and Enhanced Dielectric Properties of PVA/NiO/Ti Nanostructures for Nanoelectronic Fields

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This work aims to investigate the synthesis of PVA–NiO–Ti nanostructures and exploring their dielectric properties to use in various nanoelectronics applications. The dielectric properties of PVA–NiO–Ti nanostructures are examined in the range of frequency between 100 Hz and 2 MHz. The experimental results demonstrate that the dielectric constant, dielectric loss, and electrical conductivity of PVA are improved with increasing NiO–Ti-nanoparticles' ratio. The dielectric constant and dielectric loss of PVA–NiO–Ti nanostructures are reducing, while the electrical conductivity is rising with increasing frequency. The attained results of dielectric properties for PVA–NiO–Ti nanostructures indicate that these nanocomposites are useful for various nanoelectronics applications.

Метою цієї роботи є дослідження синтези наноструктур ПВС–NiO–Ti та вивчення їхніх діелектричних властивостей для використання в різних застосуваннях наноелектроніки. Діелектричні властивості наноструктур ПВС–NiO–Ti досліджувалися в діапазоні частот від 100 Гц до 2 МГц. Експериментальні результати показали, що діелектрична проникність, діелектричні втрати й електропровідність ПВС поліпшувалися зі збільшенням співвідношення наночастинок NiO–Ti. Діелектрична проникність і діелектричні втрати наноструктур ПВС–NiO–Ti зменшувалися, тоді як електропровідність зростала зі збільшенням частоти. Одержані результати дослідження діелектричних властивостей наноструктур ПВС–NiO–Ti вказують на те, що ці наноконізити корисні для різних застосувань у наноелектроніці.

Key words: nanocomposites, PVA–NiO–Ti, dielectric properties, conductivity.

Ключові слова: нанокompозити, ПВС–NiO–Ti, діелектричні властивості, провідність.

(Received 28 November, 2023)

1. INTRODUCTION

Hybrid (organic–inorganic) nanocomposites have recently attracted more attention for their outstanding physical and chemical properties. Hybrid nanocomposite films are used in various technological and industrial fields, including supercapacitors, optical lenses, light-emitting diodes (LED), photodetectors, solar cells, multisensors, and UV shielding [1–4]. Polymer-matrix nanocomposites, which exhibit distinct physicochemical characteristics by incorporating inorganic fillers into polymer networks, have received much attention due to their various industrial applications in drug delivery, water treatment, food industry, aeronautical and aerospace structures [5]. Metal nanoparticles (NPs) combined with polymers are greatly attractive because of the various applications offered by these materials. More development of the nanocomposites properties can be enhanced by loading nanofiller materials with aspect ratio [6]. Polyvinyl alcohol (PVA) has significant potential for groundbreaking innovations. It is a non-toxic, water-soluble, and biocompatible synthetic polymer and is mostly utilized in the pharmaceutical, packaging, and biomedical industries with various purposes. The chemical structure of PVA is simple and consists of the main structure of carbons with hydroxyl groups. Because of intra- and intermolecular hydrogen interactions, PVA samples dissolve at high temperatures (of about 80°C) and require constant stirring. Additionally, it has great film-forming properties, high chemical resistance, and a PVA surface layer that prevents nanocomposites from clumping together, which causes them to become stable and monodispersed [7]. The nanocomposites of inorganic-doped polymers or inorganic–inorganic included numerous applications in various fields involved thermal energy storage [8–16], antibacterial [17–25], optical fields [26–35], *etc.*

The present study investigates the fabrication of PVA–NiO–Ti nanostructures to utilize in various nanoelectronics applications.

2. MATERIALS AND METHODS

The samples of pure PVA and PVA–NiO–Ti nanostructures were synthesized with various contents of PVA and NiO–Ti NPs *via* cast-

ing method. The PVA of 1 gm was dissolved in 30 ml of distilled water to produce of PVA film. The NiO–Ti NPs were added to the PVA solution by contents of 1, 2 and 3 wt.% with concentration 50% NiO:50% Ti. The dielectric properties of PVA–NiO–Ti nanostructures were investigated at the frequency from 100 Hz to 2 MHz using LCR meter type (HIOKI 3532-50 LCR HI TESTER). The dielectric constant, ϵ' , is defined by Ref. [36] as follows:

$$\epsilon' = C_p/C_0, \quad (1)$$

where C_p is the matter capacitance and C_0 is the vacuum capacitor.

The dielectric loss, ϵ'' , is determined by Ref. [37] as

$$\epsilon'' = \epsilon' D, \quad (2)$$

where D is represented dispersion factor.

The A.C. conductivity is given by Ref. [38] as

$$\sigma_{A.C.} = \omega \epsilon'' \epsilon_0, \quad (3)$$

where ω is the angular frequency.

3. RESULTS AND DISCUSSION

Figures 1 and 2 confirm the variations of ϵ' and ϵ'' for PVA–NiO–Ti nanostructures with frequency, respectively. From these figures, the ϵ' includes elevated values at low frequency. The ϵ' value of dielectric constant at low frequency might be related to the electrode effect and interfacial effect of the film. The elevated value of ϵ'' for PVA–NiO–Ti nanostructures at low frequency is due to the charges' mobility. The ϵ' and ϵ'' are rising with increasing NiO–Ti nanoparticles' content; this result can be related to the rise in conductivity because of the rising density of charge carriers in polymer medium [39–55].

Figure 3 illustrates the A.C. electrical conductivity behaviour with frequency for PVA–NiO–Ti nanostructures. With decreasing frequency, the more charge is accumulated at the electrode and electrode interface that leads to reducing in the number of mobile ions and, eventually, to a drop in the conductivity at low frequencies. At high frequency, the conductivity increases with the frequency as related to the mobility of charge carriers and the hopping of ions from the infinite cluster. As a result, the ion exchange process occurs effectively in the high-frequency region. The A.C. electrical conductivity of PVA is increasing with an increase in the NiO–Ti-NPs' content; this performance is due to the improvement

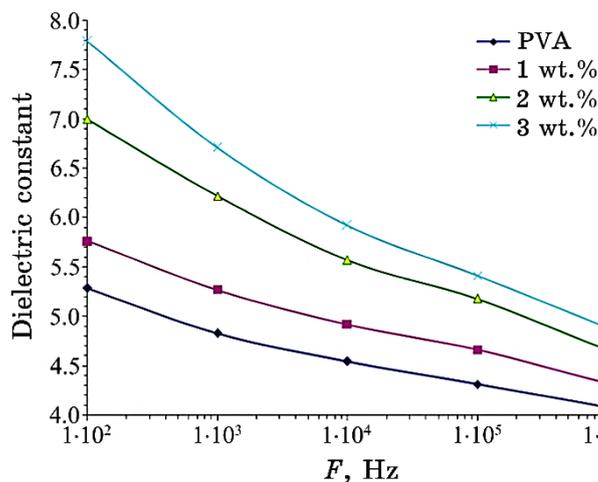


Fig. 1. Variation of ϵ' for PVA–NiO–Ti nanostructures with frequency F .

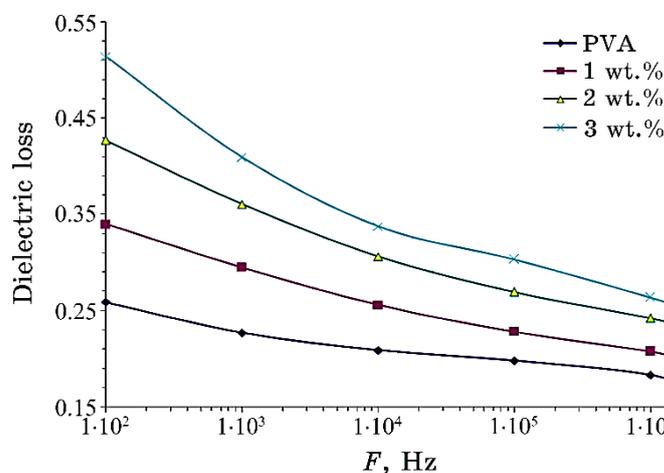


Fig. 2. Behaviour of ϵ'' for PVA–NiO–Ti nanostructures with frequency F .

of the mobility of charge ions and the larger number of charge carriers in polymer medium [56–67].

4. CONCLUSIONS

Fabrication of PVA and PVA NiO–Ti-doped nanostructures' films are investigated in this work. The dielectric properties of PVA–NiO–Ti nanostructures were studied to utilize in various nanoelectronics fields. The results demonstrate that the dielectric constant,

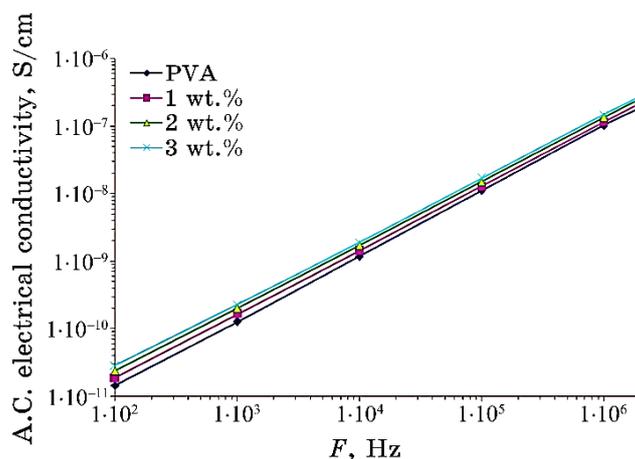


Fig. 3. Performance of A.C. electrical conductivity with frequency F for PVA–NiO–Ti nanostructures.

dielectric loss, and electrical conductivity of PVA are increased with increasing NiO–Ti-NPs' content. The dielectric constant is enhanced by 32.1% at 100 Hz, when the NiO–Ti-NPs' content of 3% is reached. The dielectric constant and dielectric loss of PVA–NiO–Ti nanostructures are decreased, while the electrical conductivity is rising with increasing frequency. The obtained results for dielectric properties of PVA–NiO–Ti nanostructures show that these nanocomposites are functional materials for deferent nanoelectronics fields.

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