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Exploring the Optical Properties of PVA–ZrO₂–Sb₂O₃ Nanostructures for Low-Cost Nanoelectronics Fields

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The present work goals are to fabricate the PVA–ZrO₂–Sb₂O₃ nanostructures to use in low-cost nanoelectronics applications. The optical properties of PVA–ZrO₂–Sb₂O₃ nanostructures are examined at wavelength ranged from 200 nm to 800 nm. The results confirm that the absorbance and absorption coefficient of PVA are increased with rising in the ZrO₂–Sb₂O₃-nanoparticles' content. The transmittance and energy gap are decreased with increasing in the ZrO₂–Sb₂O₃-nanoparticles' content. The results indicate that the PVA–ZrO₂–Sb₂O₃ nanostructures can be considered as a key for various nanoelectronics fields.

Метою даної роботи є виготовлення наноструктур полівініловий спирт (ПВС)–ZrO₂–Sb₂O₃ для використання в недорогих застосуваннях наноелектроніки. Досліджено оптичні властивості наноструктур ПВС–ZrO₂–Sb₂O₃ в діапазоні довжин хвиль від 200 нм до 800 нм. Результати підтверджують, що абсорбція та коефіцієнт поглинання ПВС зростають зі збільшенням вмісту наночастинок (НЧ) ZrO₂–Sb₂O₃. Зі збільшенням вмісту НЧ ZrO₂–Sb₂O₃ коефіцієнт пропускання й енергетична щільність зменшуються. Результати вказують на те, що наноструктури ПВС–ZrO₂–Sb₂O₃ можна вважати ключовими для різних галузей наноелектроніки.

Key words: PVA, ZrO₂–Sb₂O₃, nanocomposites, energy gap, optical properties.

Ключові слова: полівініловий спирт, ZrO₂–Sb₂O₃, нанокompозити, енергетична щільність, оптичні властивості.

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1. INTRODUCTION

Electronics has been undergoing a disruptive evolution, investing in lightweight, soft and flexible devices instead of heavy, bulky, and rigid devices. Thus, flexible electronics is a fast-growing field that promises to develop new commercial products such as displays, solar cells, and biomedical sensors. These devices can be incorporated into clothing and other everyday items that ensure they revolutionize our daily lives. In this way, the electronic industry opens up new opportunities, and endless manufacturing advances in thin-film materials and devices will make flexible electronics ubiquitous [1]. Semiconductor materials have yielded valuable information beyond the most advanced knowledge in the technological field in recent years. These materials have essential widespread applications, including–VI semiconductor materials, such as solar cells, light emitting devices [2].

The surface of metal oxides is a key factor for effective interaction with target molecules, however, reducing the size of metal oxide particles to the nanoscale, increases the active surface area and induces a new effect due to quantum confinement such as band gap widening, UV-absorption, room temperature, and photoluminescence [3].

Polyvinyl alcohol (PVA) is a semicrystalline polymer having a high dielectric strength, good thermostability, high mechanical strength, chemical resistance, excellent film forming properties and optical transparency. The existence of the hydroxyl group, ($-\text{OH}$), gives PVA special properties due to the strong $-\text{OH}$ interaction between intra and intermolecular polymer chains [4].

Antimony trioxide nanoparticles (NPs; nano- Sb_2O_3) have been widely concerned because of good synergistic fire-retardant effect between nano- Sb_2O_3 and halogen flame-retardants [5].

Nanosize zirconia has attracted much attention due to its specific optical and electrical properties as well as other potential applications in transparent optical devices, electrochemical capacitor electrodes, oxygen sensors, fuel cells, catalysts and advanced ceramics [6].

The nanocomposites compose of two or more materials have huge applications in various fields like electronics and optoelectronics [7–21], optical fields [22–31], radiation shielding and bioenvironmental [32–37], sensors [38–40], energy storage [41–43] and antibacterial [44–50]. This work deals with fabrication and optical properties of PVA- ZrO_2 - Sb_2O_3 nanostructures to use in nanoelectronics fields.

2. MATERIALS AND METHODS

Films of PVA-ZrO₂-Sb₂O₃ nanostructures were fabricated by using casting technique. The 0.5 gm of PVA was dissolved in 30 ml of distilled water by using magnetic stirrer to mix the polymer for 1 hour to obtain more homogeneous solution. Then, ZrO₂ and Sb₂O₃ nanoparticles were added to PVA solution with constant ratio 1:1 and contents of 2.2%, 4.4% and 6.6%. The optical properties of PVA-ZrO₂-Sb₂O₃-nanostructures' films were examined using the double beam spectrophotometer (Shimadzu, UV-1800Å) at wavelength ranged from 200 nm to 800 nm.

The absorption coefficient (α) is given by [51] as follows:

$$\alpha = 2.303(A/t), \quad (1)$$

where A is the absorbance and t is the thickness of films.

The energy gap is determined by means of equation [52]

$$\alpha h\nu = C(h\nu - E_g)^r, \quad (2)$$

where C is the constant; $h\nu$ is the energy of photon; E_g is the energy gap, and $r=2$ and 3 for allowed and forbidden indirect transitions.

3. RESULTS AND DISCUSSION

Figures 1 and 2 show the performances of absorbance and transmittance of PVA-ZrO₂-Sb₂O₃ nanostructures with wavelength respectively. As shown in these figures, the absorbance is high at low wavelength (UV-region) due to the high energy at this region. The absorbance reduces and transmittance increases with increasing wavelength. The absorption of PVA rises, while the transmittance reduces, when the ZrO₂-Sb₂O₃ NPs content rises, which may be due to increase in the number of charge carriers. This enhancement in the absorption of PVA by doping with ZrO₂-Sb₂O₃ NPs can due to improve of other intermolecular bonds between cations and anions creating defects over the entire polymer medium. The reduced transmittance is due to the rise in scattering processes of the incident photons by denser NPs' filling the polymer matrix [53–65].

Figure 3 demonstrates the variation of absorption coefficient for PVA-ZrO₂-Sb₂O₃ nanostructures with photon energy. From this figure, the values of absorption coefficient are less than 10^4 cm^{-1} that indicates to the indirect transition.

Figures 4 and 5 illustrate the energy gap values for the allowed and forbidden transitions of PVA-ZrO₂-Sb₂O₃ nanostructures. The

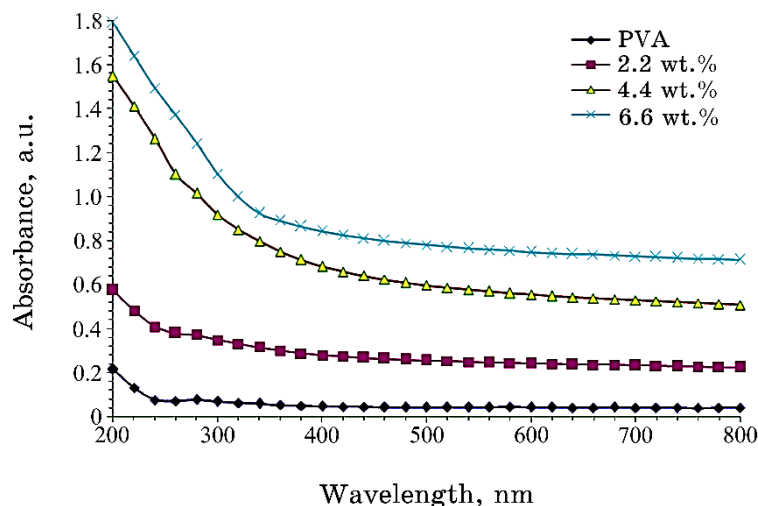


Fig. 1. Performance of absorbance for PVA-ZrO₂-Sb₂O₃ nanostructures with wavelength.

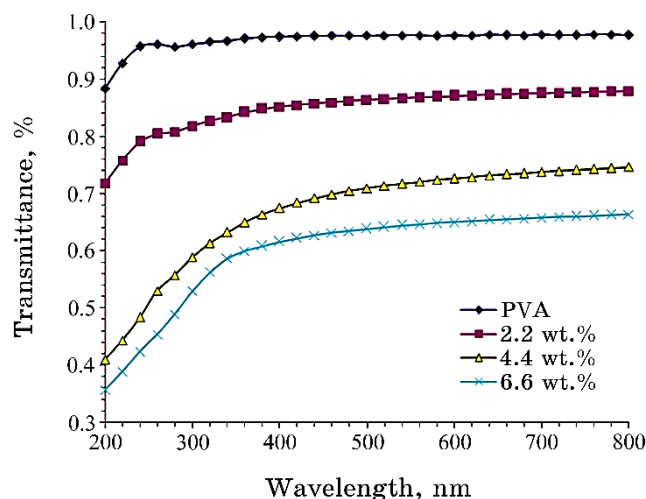


Fig. 2. Transmittance variation of PVA-ZrO₂-Sb₂O₃ nanostructures with wavelength.

energy gap of PVA is reduced with rising ZrO₂-Sb₂O₃ content; this behaviour can be due to the creation of new energy levels, which create the energy band gap. The reduced energy gap selects that the charge transfer complexes presence relate the form of defects in the polymer matrix. These defects create the localized levels in the energy gap [66–72].

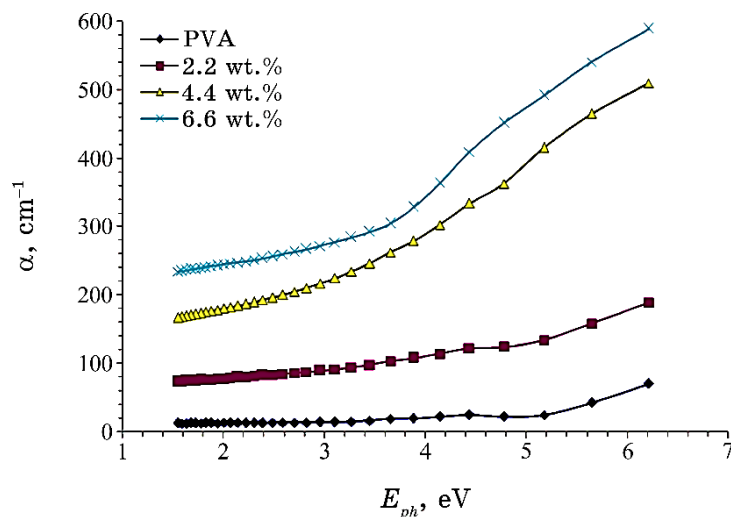


Fig. 3. Variation of absorption coefficient for PVA-ZrO₂-Sb₂O₃ nanostructures with photon energy.

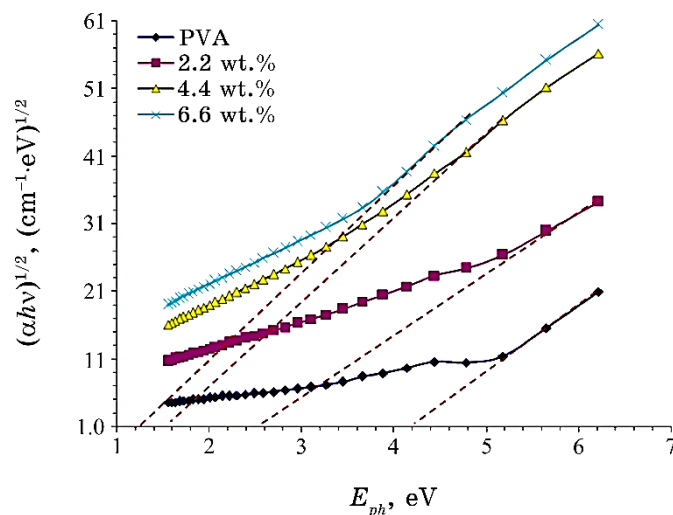


Fig. 4. Energy gap values for allowed transition for PVA-ZrO₂-Sb₂O₃ nanostructures.

4. CONCLUSION

This paper includes synthesis of PVA-ZrO₂-Sb₂O₃ nanostructures to use in low-cost nanoelectronics fields. The experimental results showed that the absorbance and absorption coefficient of PVA are

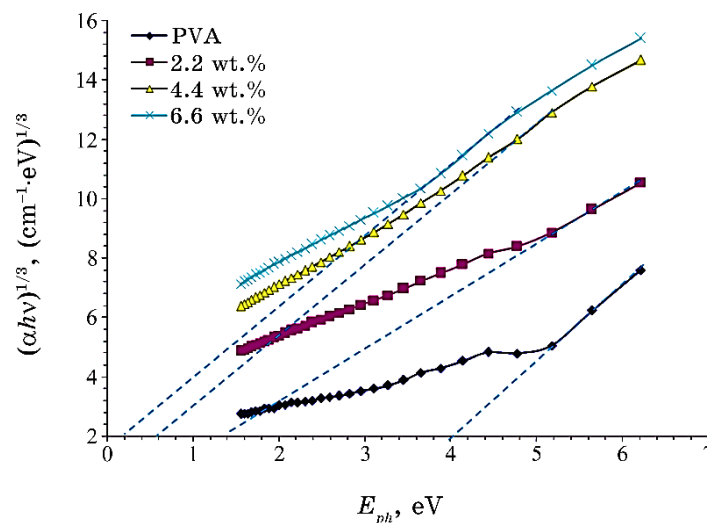


Fig. 5. Energy gap values for forbidden transition for PVA-ZrO₂-Sb₂O₃ nanostructures

increased with increasing of ZrO₂-Sb₂O₃-NPs' content. The transmittance and energy gap are decreased with increasing of the ZrO₂-Sb₂O₃-NPs' content. Finally, the results demonstrated that the PVA-ZrO₂-Sb₂O₃ nanostructures can be considered as promising materials for nanoelectronics fields.

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