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## **Investigating the A.C. Electrical Properties of PVA/PEG/CeO<sub>2</sub> Nanocomposites for Dielectric Applications**

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Films of PVA/PEG/CeO<sub>2</sub> nanocomposites are fabricated to be used in different electronic applications. The A.C. electrical properties of PVA/PEG/CeO<sub>2</sub> nanocomposites are studied. The dielectric properties are examined at frequencies from 100 Hz to 5 MHz. The experimental results indicate that the A.C. electrical properties of PVA/PEG/CeO<sub>2</sub> nanocomposites are enhanced with increase in the CeO<sub>2</sub> nanoparticles' content. The dielectric parameters are changed with rise in the frequency.

Плівки нанокompозитів PVA/PEG/CeO<sub>2</sub> (полівініловий спирт–поліетиленгліколь–наночастинки CeO<sub>2</sub>) виготовляються для використання в різних електронних застосуваннях. Вивчено електричні властивості нанокompозитів PVA/PEG/CeO<sub>2</sub>. Діелектричні властивості досліджуються на частотах від 100 Гц до 5 МГц. Експериментальні результати показують, що електричні властивості нанокompозитів PVA/PEG/CeO<sub>2</sub> підвищуються зі збільшенням вмісту наночастинок CeO<sub>2</sub>. Параметри діелектрика змінюються з підвищенням частоти.

**Key words:** PVA/PEG, CeO<sub>2</sub> nanoparticles, nanocomposites, electrical properties, conductivity.

**Ключові слова:** полівініловий спирт/поліетиленгліколь, наночастинки CeO<sub>2</sub>, нанокompозити, електричні властивості, провідність.

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## 1. INTRODUCTION

Polymer composites are polymers containing modifiers that are utilized as passive and active layers in optoelectronic devices, such as solar cells, light-emitting diodes (LED), optical waveguides materials, and photochromic materials [1].

Polyvinyl alcohol (PVA) is a synthetic, non-toxic biocompatible, semi-crystalline polymer having good transparency, high dielectric strength, and fast charge transfer at electrode–electrolyte interface. Due to its excellent optical and physical properties, PVA is successfully used in a wide range of industrial field [2]. PVA being a hydrophilic–organic polymer has fascinated the interest of materials scientists due to its variety of applications in food packaging, humidity sensors, thin-film transistors, fuel cells, and so on [3]. Moreover, PVA has a carbon-chain backbone with hydroxyl group that is a source of hydrogen bonding to improve the development of polymer complexes.

Furthermore, polyethylene glycol (PEG) has significant features such as superb water solubility, being good protein adsorption resistance and low toxicity; therefore, it is used in several applications such as a coating material for biotechnical applications [4].

CeO<sub>2</sub> is one of the most attractive rare-earth metal oxides. It has several applications in the field of corrosion prevention, electrochemical cells, electromagnetic shielding, thermal coatings, optical and photoelectrical properties. Mostly, the rare-earth elements exist in trivalent state, but cerium exists in both trivalent (+3) state and tetravalent (+4) one. Because of its ability to alter oxidation state, it has the several applications such as catalytic converters and solid oxide fuel cells. In addition, CeO<sub>2</sub> has a great attention because of its unique features like nontoxicity, biocompatibility, oxygen storage capability, optical and thermal properties, which have significant applications in solar cells, gas sensors and biosensors [5].

This paper deals with preparation of PVA/PEG/CeO<sub>2</sub> nanocomposites for films to be employed in different electronic applications.

## 2. MATERIALS AND METHODS

Films PVA/PEG/CeO<sub>2</sub> nanocomposites were synthesized by using casting technique. The PVA/PEG blend was prepared with concentrations of 81 wt.% PVA and 19 wt.% PEG by dissolving of 1 gm of polymers in distilled water (30 ml). The CeO<sub>2</sub> nanoparticles (NPs)

were added to the polymer blend with contents of 1.5 wt.%, 3 wt.%, and 4.5 wt.%. The A.C. electrical properties of PVA/PEG/CeO<sub>2</sub> nanocomposites were measured within the frequency range 100 Hz–5·10<sup>6</sup> Hz by LCR meter (HIOKI 3532-50 LCR HI TESTER).

The dielectric constant ( $\epsilon'$ ) of nanocomposite films was calculated by [6]:

$$\epsilon' = C_p d / \epsilon_0, \quad (1)$$

where  $C_p$  is matter capacitance,  $d$  is the thickness,  $A$  is the area (in cm<sup>2</sup>).

Dielectric loss ( $\epsilon''$ ) was determined by [6]:

$$\epsilon'' = \epsilon' D, \quad (2)$$

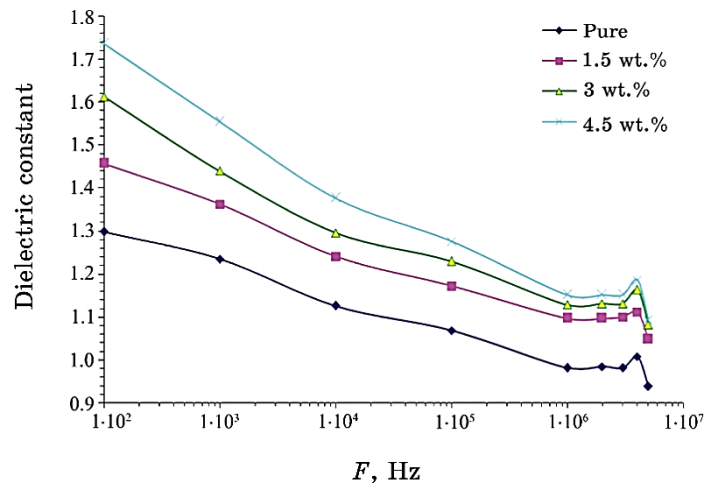
where  $D$  is the dispersion factor.

The A.C. electrical conductivity was found by [7]:

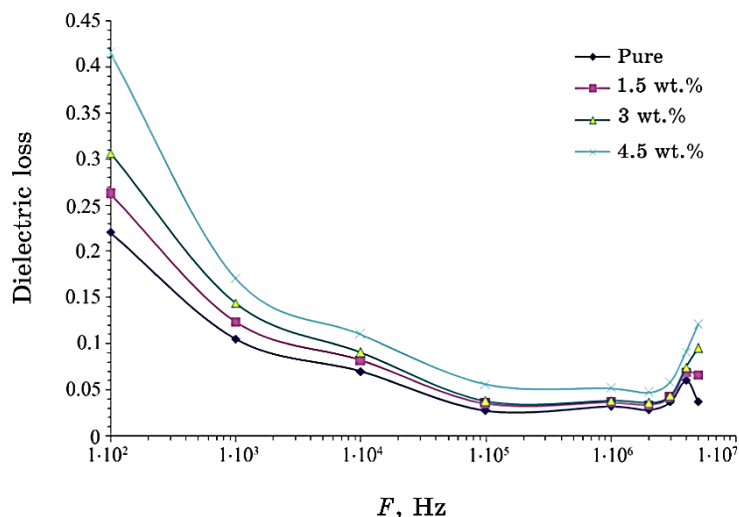
$$\sigma_{A.C.} = \omega \epsilon' D \epsilon_0. \quad (3)$$

### 3. RESULTS AND DISCUSSION

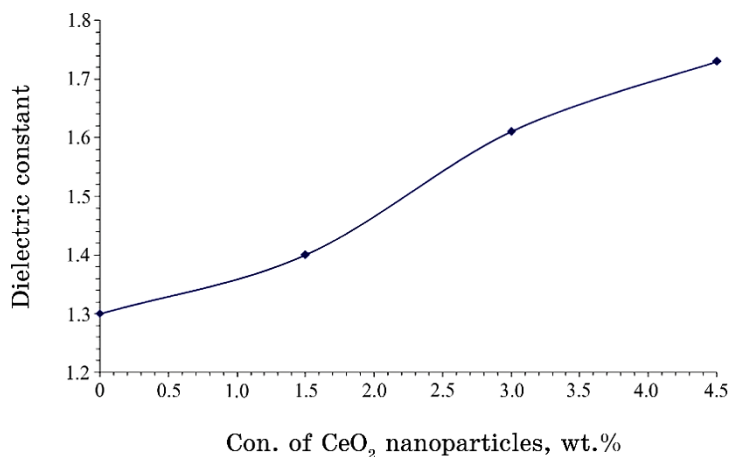
Figures 1–4 show the variations of dielectric constant and dielectric loss of PVA/PEG/CeO<sub>2</sub> nanocomposites with frequency and CeO<sub>2</sub> NPs' concentration, respectively.



**Fig. 1.** Variation of dielectric constant of PVA/PEG/CeO<sub>2</sub> nanocomposites with frequency.



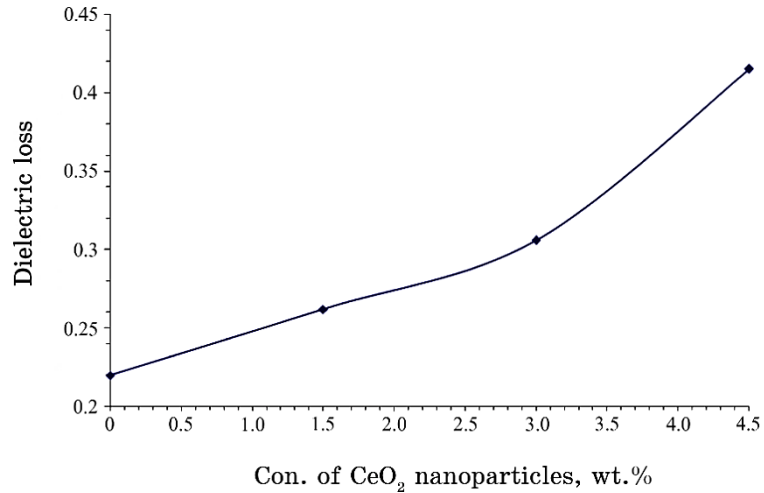
**Fig. 2.** Variation of dielectric loss of PVA/PEG/CeO<sub>2</sub> nanocomposites with frequency.



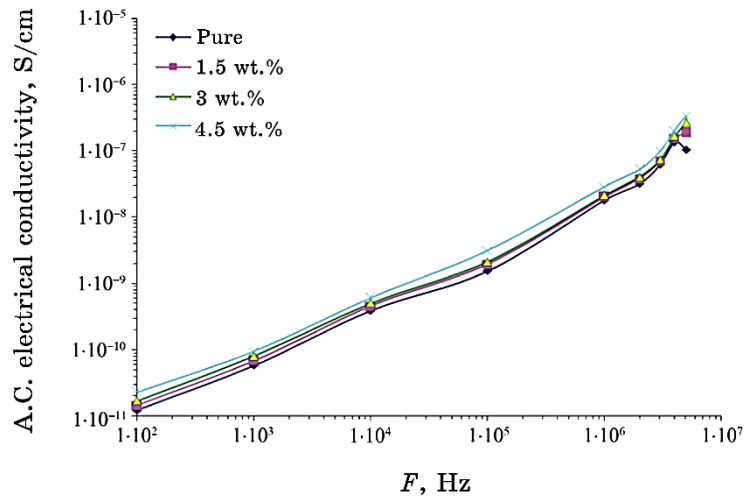
**Fig. 3.** Variation of dielectric constant of PVA/PEG blend with CeO<sub>2</sub> NPs' concentration.

The dielectric constant and dielectric loss are decreased with increase in the frequency. This is because the effect of space charge polarization has decreased. Because of the charge accumulated due to the polarization of the PVA/PEG/3 wt.% CeO<sub>2</sub> NPs' interfaces, more charges are generated, resulting in an increase in the number of charge carriers accumulated at the sample-electrode interface [8–14].

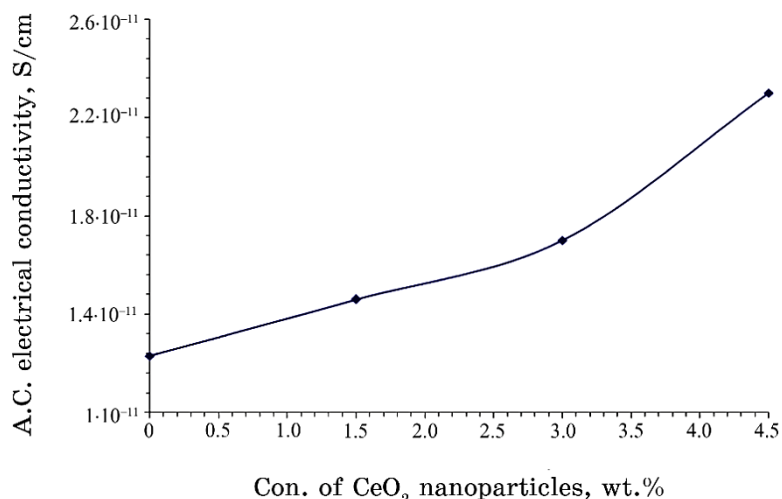
Figure 5 and Figure 6 represent the behaviour of A.C. electrical conductivity for PVA/PEG/CeO<sub>2</sub> nanocomposites with frequency and CeO<sub>2</sub> NPs' concentration, respectively. The conductivity increases with increase in frequency as well as CeO<sub>2</sub> NPs' concentration. This increase is attributed to the formation of excess charge carriers developed in the polymer matrix [15–24].



**Fig. 4.** Variation of dielectric loss of PVA/PEG blend with CeO<sub>2</sub> NPs' concentration.



**Fig. 5.** Behaviour of A.C. electrical conductivity for PVA/PEG/CeO<sub>2</sub> nanocomposites with frequency.



**Fig. 6.** Behaviour of A.C. electrical conductivity for PVA/PEG blend with CeO<sub>2</sub> NPs' concentration.

#### 4. CONCLUSIONS

The current work includes studying the dielectric characteristics of PVA/PEG/CeO<sub>2</sub> nanocomposites to use for various electrical devices. The results indicated to enhance the dielectric properties of PVA/PEG with adding of the CeO<sub>2</sub> NPs' concentration. The dielectric constant, dielectric loss and A.C. electrical conductivity are changed with the increase in frequency.

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